PREPARATION OF (E) $-\alpha$, β -UNSATURATED AROMATIC ALDEHYDES BY HIGH STEREOSELECTIVE OLEFINFORMYLATION.

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Abstract: Treatment of acetaldehyde N-ter-butylimine by two equivalents of LDA, followed by trimethylsilyl chloride trapping affords the α, α -disilylated imine; the latter reacts with aromatic aldehydes producing α, β -ethylenic aldehydes in excellent yields and with very high stereoselectivity in favour of the E isomer.

The conversion of an aldehyde into its vinylogue¹ is an important organic synthetic reaction since the α,β -unsaturated carbonyl compounds thus obtained are often used as intermediates in the preparation of natural compounds². The most known methods refer to the crotonisation, to the Wittig reaction or to its variations³.

During the last years, the authors have directed their efforts towards the utilization of silylated reagents in order to create a carbon-carbon double bond by the Peterson olefination⁴. Thus, Corey and Coll.⁵ employed the lithiated derivatives of α -silyl aldimines whereas Duhamel and Coll.⁶ carried out a similar process using α -silylated acetaldehyde.

Generally, the authors obtain the aldehydes of E configuration whatever method applied. However, small quantities of the Z isomer contaminate the main product and chromatographic separation is necessary.

In connection's with our work on silicon derivatives concerning a new preparation of Tigason^R and its analogues's, we describe in this report a new reagent permitting the transformation of an aromatic aldehyde into its vinylogue with very high stereoselectivity in favour of the E configuration.

This makes use of the disilylated imine 2, obtained as indicated below.

The NMR spectrum of $\underline{2}$ confirms the formula proposed and excludes an enaminate structure. The intermediate $\underline{3}$ is not isolable. However, $\underline{4}$ has been easily isolated and characterised by NMR. The yields indicated are based on starting aldehydes without isolation of $\underline{4}$.

The E and Z isomers of α,β -unsaturated aldehydes are easily distinguished by the NMR signal of the formyl group¹⁰. In order to ensure the steric purity of our compounds, we prepared the <u>5aE/5aZ</u> 96/4 mixture from authentic samples. <u>5aZ</u> is easily revealed and quantified (¹H-RMN 60 MHz; solution 30 %).

^{*}The indicated yields refer to crude products obtained after evaporation of the solvent. The $^1\text{H-NMR}$ 60 MHz spectrum (concentrated solutions \sim 30 %) does not show any impurity.

The aliphatic aldehydes lead to the same reaction; however we have not managed yet to optimize the yields. Ketones are enclised and do not react further.

If the C-Si bonds are activated by tetrabutylammonium fluoride, mixture (50/50) of the two isomers was obtained. Thus, this reaction could be used for the synthesis of α,β -unsaturated aromatic aldehydes of Z configuration.

Me₃Si

NtBu

$$+ Ph-CHO$$
 $\xrightarrow{BuN_4^+F^- \text{ cat.}}$
 $+ Ph-CHO$
 $+ Ph$
 $+ Ph$
 $+ Ph-CHO$
 $+ Ph$
 $+ P$

EXPERIMENTAL ILLUSTRATION

 α, α -Bis(trimethylsilyl) ter-butylacetaldimine 2. LDA is prepared by attack of the amine by lithium in the presence of an electron acceptor (α -methylstyrolene) in ether-THF 50/50. The imine is deprotonated 11 at -70°/-60°C during 18 h. One equivalent of ClSiMe3 is added and the mixture is stirred for 5 h at -60°C. Then a second equivalent of ClSiMe3 is added at -60°C, the solution is kept at the same temperature for 1 h and then allowed to warm up to room temperature. The precipitate is filtered off, the solvent removed, the solid is filtered off again and the product distilled. 97-98°C, 12 torr.

 $^{1}H\text{-NMR}$ 60 MHz ; CCl₄ ; TMS ; $\delta(ppm)$; J(Hz) : 0.04, 18 H, s ; 1.10, 9 H, s ; 1.41, 1 H, d, 7.7 ; 7.89, 1 H, d, 7.7.

p-Methoxycinnamaldehyde $\underline{5c}$. Reagent $\underline{2}$ (0,02 mole), $\operatorname{ZnBr_2}$ cat. (10 %) and p-methoxybenzaldehyde in deficient quantity (0,018 mole) are mixed in 20 ml THF and left aside for 6 h at 10°C, then overnight at 20°C. The imine $\underline{4}$ is extracted as a salt by a cooled dilute aqueous solution of $\operatorname{H_2SO_4}$, then freed by NaOH and extracted with ether. This ether solution is stirred with an aqueous solution of $\operatorname{ZnCl_2}$ (5 %) for 6 h at 10°C, then left aside overnight at 20°C. The precipitate of $\operatorname{Zn(OH)_2}$ is filtered off, the organic layer washed by an aqueous diluted solution of $\operatorname{H_2SO_4}$, then by water and dried on MgSO₄. By removing the solvent in vacuo, $\underline{5c}$ crystallizes.

REFERENCES AND FOOTNOTES

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- 11 The sequence is probably as follows:

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